

Dense Phase CO₂ Pipeline Case Study Using ProMax Properties

Input data:

$$\text{kmole} := 1000 \text{mole} \quad \text{lbmole} := 453.4 \text{mole}$$

$$\text{MW} := 44 \frac{\text{lb}}{\text{lbmole}} \quad cP := \frac{\text{poise}}{100} \quad C := 5.7 \cdot 10^{-4} \frac{\text{MPa}}{\text{K}} \quad R_g := 8.314 \cdot \frac{\text{Pa} \cdot \text{m}^3}{\text{mole} \cdot \text{K}}$$

$$Q_S := 160 \cdot 10^6 \frac{\text{SCF}}{\text{D}}$$

$$Q_{\text{Std}} := Q_S \frac{\text{ft}^3}{\text{day}}$$

CO₂ Critical Properties $T_C := 304 \text{K}$ $P_C := 7.38 \text{MPa}$ $\text{MW} = 44.019 \frac{\text{kg}}{\text{kmole}}$

$$\text{mass} := Q_S \frac{\text{ft}^3}{\text{day}} \cdot \frac{\text{lbmole}}{379.5 \cdot \text{ft}^3} \cdot \text{MW}$$

$$\text{mass} = 97.39 \frac{\text{kg}}{\text{s}}$$

$$\text{mass} = 214.707 \frac{\text{lb}}{\text{s}}$$

$$\text{MoleRate} := Q_{\text{Std}} \left(\frac{\text{lbmole}}{379.5 \cdot \text{ft}^3} \right) \quad \text{MoleRate} = 1.757 \times 10^4 \frac{\text{lbmole}}{\text{hr}} \quad \text{MoleRate} = 7.965 \times 10^3 \frac{\text{kmole}}{\text{hr}}$$

$$\text{Length} := 100 \cdot \text{mile} \quad \text{Length} = 160.934 \cdot \text{km} \quad \text{Dia} := 395 \text{mm} \quad \text{Dia} = 15.551 \text{ in}$$

$$\text{Area} := \frac{\pi}{4} \cdot \text{Dia}^2 \quad \text{Area} = 0.12254 \text{ m}^2 \quad \text{Area} = 1.319 \cdot \text{ft}^2$$

Standard Condition: $P_S := 101.325 \cdot 10^3 \text{Pa}$ $T_S := (15 + 273) \text{K}$

$$\rho_S := \frac{P_S \cdot \text{MW}}{R_g \cdot T_S} \quad \rho_S = 1.863 \cdot \frac{\text{kg}}{\text{m}^3}$$

$$Q_{\text{Std}} := \text{MoleRate} \cdot \left(23.64 \frac{\text{m}^3}{\text{kmole}} \right)$$

$$Q_{\text{Std}} = 52.303 \cdot \frac{\text{m}^3}{\text{s}} \quad Q_{\text{Std}} = 4.519 \times 10^6 \frac{\text{m}^3}{\text{day}}$$

$$T := (40 + 273) \text{K} \quad T = 313 \text{ K} \quad T = 563.4 \text{ R}$$

Solution for one segment

$$P_{\text{in}} := 140 \text{bar} \quad P_{\text{in}} = 14 \cdot \text{MPa} \quad P_1 := P_{\text{in}} \quad P_{\text{out}} := 9.815 \text{MPa} \quad (\text{Assumed Value})$$

$$P_{\text{Avg}} := \frac{P_{\text{in}} + P_{\text{out}}}{2} \quad P_{\text{Avg}} = 119.075 \cdot \text{bar} \quad P_{\text{in}} = 2.0305 \times 10^3 \cdot \text{psi}$$

$$P_{\text{Avg}} := \frac{P_{\text{Avg}}}{\text{MPa}} \quad P_{\text{Avg}} = 11.908 \quad P_{\text{Avg}} = 11.908 \cdot \text{MPa}$$

Density and Viscosity Correlation Coefficients

$$DA := \begin{pmatrix} -4722.2000 \\ 1217.5000 \\ -94.0780 \\ 2.4694 \end{pmatrix} \quad VA := \begin{pmatrix} -0.180207558 \\ 4.5871475418E-02 \\ -3.1836754072E-03 \\ 7.7928525288E-05 \end{pmatrix}$$

$$\rho := \left(DA_4 \cdot P_{\text{Avg}}^3 + DA_3 \cdot P_{\text{Avg}}^2 + DA_2 \cdot P_{\text{Avg}} + DA_1 \right) \cdot \frac{\text{kg}}{\text{m}^3}$$

$$\rho = 605.202 \frac{\text{kg}}{\text{m}^3} \quad \rho = 37.782 \cdot \frac{\text{lb}}{\text{ft}^3}$$

$$\text{Vel} := \frac{\text{mass}}{\text{Area} \cdot \rho} \quad \text{Vel} = 1.313 \frac{\text{m}}{\text{s}} \quad \text{Vel} = 4.308 \frac{\text{ft}}{\text{s}} \quad Z := \frac{P_{\text{Avg}} \cdot \text{MW}}{\rho \cdot R_g \cdot T} \quad Z = 0.333$$

$$\mu := \left(VA_4 \cdot P_{\text{Avg}}^3 + VA_3 \cdot P_{\text{Avg}}^2 + VA_2 \cdot P_{\text{Avg}} + VA_1 \right) \cdot cP \quad \mu = 0.046 \cdot cP$$

$\mu = 4.617 \times 10^{-5} \frac{\text{kg}}{\text{m} \cdot \text{s}}$

$\mu = 3.102 \times 10^{-5} \frac{\text{lb}}{\text{ft} \cdot \text{s}}$

Calculate the Reynolds Number

$$Re := \frac{\rho \cdot (\text{Vel}) \cdot (\text{Dia})}{\mu}$$

$\varepsilon/D := 0.001$

$Re = 6.8 \times 10^6$

Roughness factor or relative roughness

Calculate the friction factor using Colburn formula by trial and error

$f := 0.02$ Use this value as initial guess

Given Find friction factor by find option of MathCad

$$\frac{1}{\sqrt{f}} = -2 \log \left(\frac{\varepsilon/D}{3.7} + \frac{2.51}{Re \cdot \sqrt{f}} \right) \quad f := \text{Find}(f) \quad f = 0.0197 \quad \text{This is calculated value}$$

Liquid Phase Pressure Drop Calculations

$$\Delta P_{\text{Liquid}} := f \cdot \frac{\text{Length}}{\text{Dia}} \cdot \frac{\text{Vel}^2}{2} \cdot \rho$$

$\Delta P_{\text{Liquid}} = 4.185 \cdot \text{MPa}$

$\Delta P_{\text{Liquid}} = 606.9 \cdot \text{psi}$

$$P_{\text{LiqOut}} := P_{\text{in}} - \Delta P_{\text{Liquid}}$$

$P_{\text{LiqOut}} = 9.815 \cdot \text{MPa}$

$P_{\text{LiqOut}} = 1423.6 \cdot \text{psi}$

Gas Phase Pressure Drop Calculations

$$P_{\text{out}} := \sqrt{P_{\text{in}}^2 - C \cdot f \cdot \text{Length} \cdot Z \cdot T \cdot \rho_S \cdot \frac{Q_{\text{Std}}^2}{\text{Dia}^5}}$$

$P_{\text{out}} = 9.815 \cdot \text{MPa}$

$P_{\text{out}} = 1423.5 \cdot \text{psi}$

$$\Delta P_{\text{Gas}} := P_{\text{in}} - P_{\text{out}}$$

$\Delta P_{\text{Gas}} = 4.185 \cdot \text{MPa}$

$\Delta P_{\text{Gas}} = 607 \cdot \text{psi}$